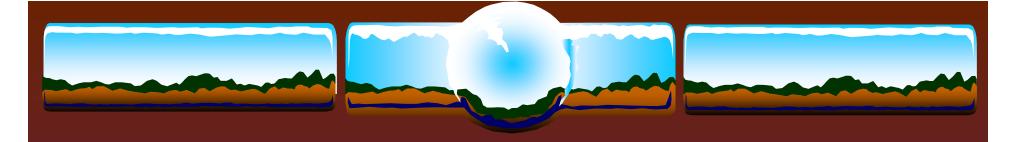


Development and Application of a Solid Phase Extraction and Micro-Liquid Chromatography-Electrospray/Ion Trap Mass Spectrometry Method for Detecting Pharmaceuticals in Natural Waters

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Development and Application of a Solid Phase Extraction and Micro-Liquid Chromatography-Electrospray/Ion trap Mass Spectrometry Method for Detecting Pharmaceuticals in Natural Waters

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U.S. EPA Notice

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- Information on exposure is essential for developing a quantitative health and environmental risk assessment of pharmaceuticals and personal care products (PPCPs) in the environment.
- Inherent to risk assessment is the development of state-of-the-art tools used to detect and monitor PPCPs in a variety of environmental media.

Objectives

To develop an analytical method that can detect non-volatile, polar, water-soluble pharmaceuticals in natural waters at levels that could be environmentally significant (at concentrations less than parts per billion, ppb).

Experimental

Aqueous Extractions

- Aqueous samples are pH adjusted to approximately pH 2.5 with 300 μL of HCl (12 N) per L of sample
- ❖ Accuprep 7000TM manifold and 48-mm nu*phase SPE C₁₈ discs (CPI International)
- Extracted with four 10-mL volumes of 99% methanol/1% acetic acid
- Evaporated to 0.1 mL using TurboVap® II (Zymark Corporation, Hopkinton, MA, USA)



Chromatography

micro-LC (liquid chromatography) columns packed in-house:

160-μm i.d. (360 -μm o.d.) x 30-cm fused silica columns (Polymicro Technologies, Phoenix, AZ). Packed with 10 to 12 cm of 5 -μm ODS-Hypersil (Shandon, Astmoor, England)

mobile phase (isocratic):

80/20/1% - methanol:water:acetic acid

flow rates: 4 to 6 µL/min

Experimental

Mass Spectrometry

- ❖ Finnigan LCQ™, configured with an electrospray (ES) ion source:
 - positive ion mode
 - ES needle: 4.8 kV
 - 180 to 800 amu (full-scan mode, in 3 microscans with an ion injection of 200ms

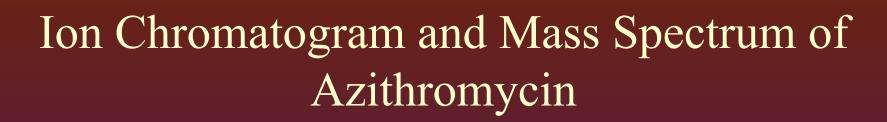
Targeted Analytes

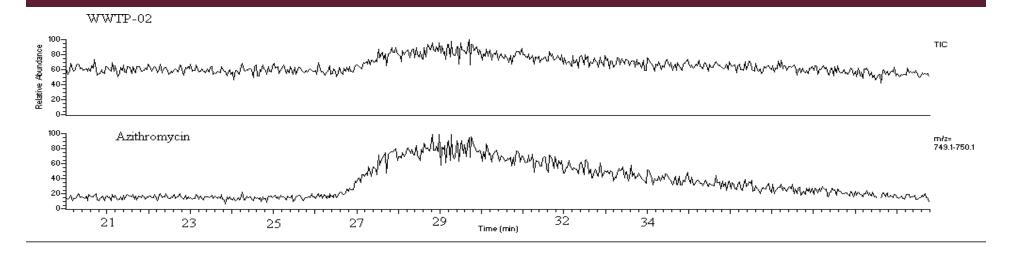
Compound	Molecular	Molecular
(# on RxList – 1999	Weight	Formula
http://www.rxlist.com/top200.htm)	Amu	
Caffeine (na)	194.19	$C_8H_{10}N_4O_2$
Fluoxetine (10)	309.33	C ₁₇ H ₁₈ F ₃ NO
Omeprazole (4)	345.42	$C_{17}H_{19}N_3O_3S$
Azithromycin (16)	749.0	$C_{38}H_{72}N_2O_{12}$
Levothyroxine Na ⁺ (2)	798.86	C ₁₅ H ₁₀ I ₄ NNaO ₄

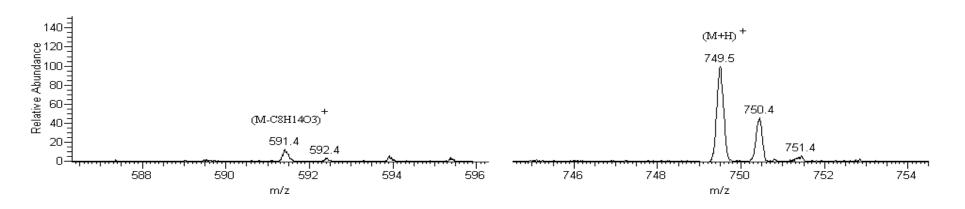
LODs* of Targeted Analytes

(*limits of detection)

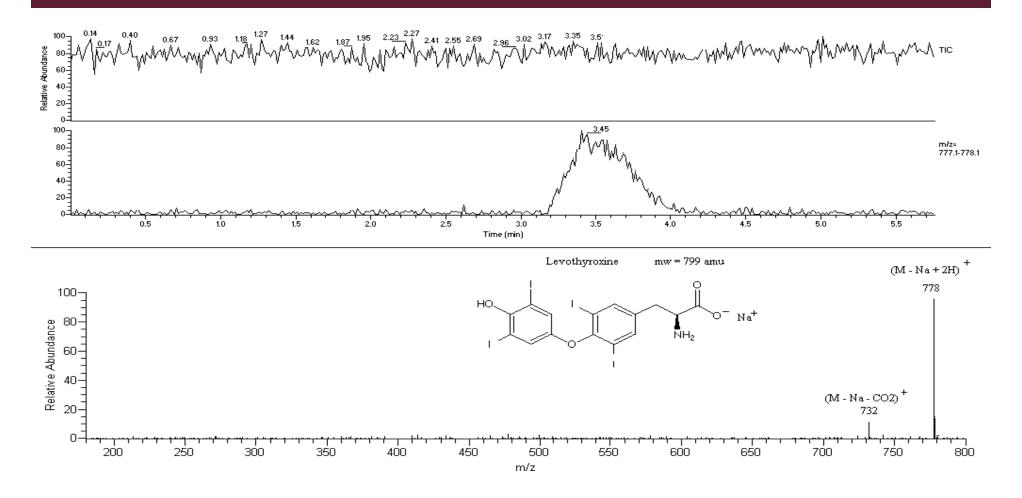
Analyte	LOD
	ng
Caffeine	3
Azithromycin	4
Levothyroxine	1
Omeprazole	1
Fluoxetine	6

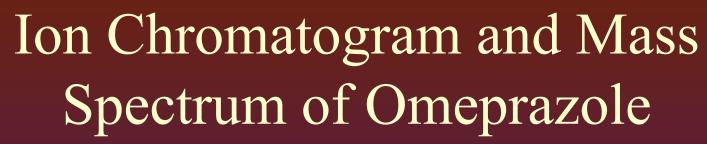


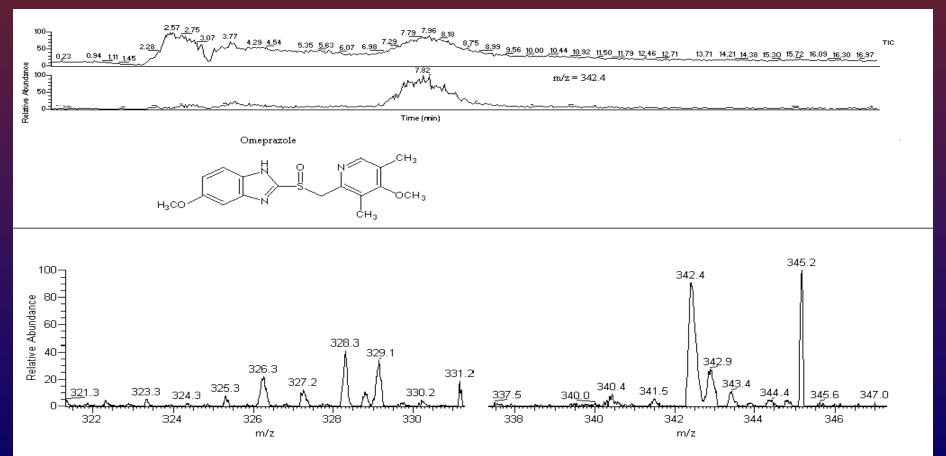




Ion Chromatogram and Mass Spectrum of Levothyroxine





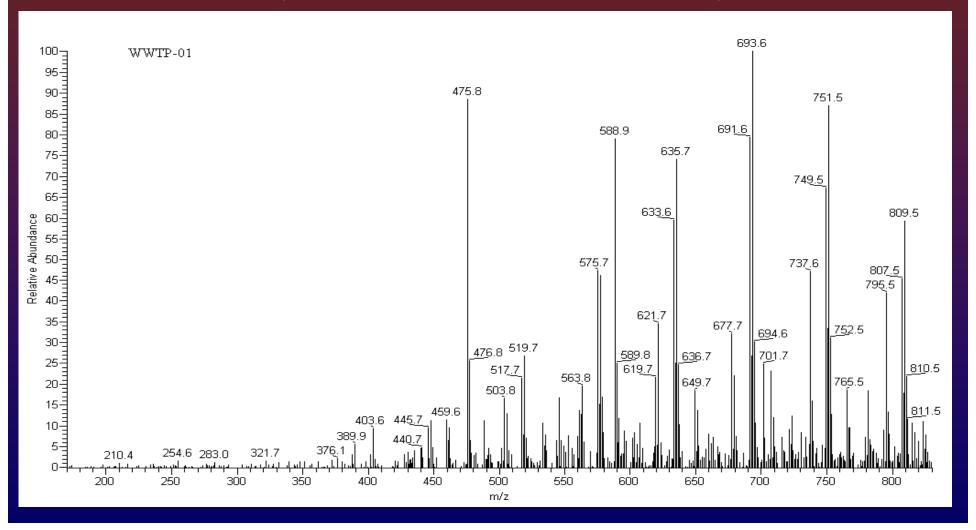


Tentatively Identified Compounds

- Other non-targeted analytes were also detected in the WWTP samples.
- For example: Polyoxypropylene glycolates (PPGs)
 and Polyoxyethylene glycolates (PEGs)
 - PPGs are widely used in the cosmetics industry as thickeners and in shampoo formulations
 - PEGs are widely used as additives in the food industry

Tentatively Identified Compounds

(a mixture of PPGs and PEGs)





(WWTP = wastewater treatment plant)

Analytes	WWTP-01	WWTP-02
	Flow = 11 mgd	Flow = 35 mgd
	< 80,000 pop.	< 250,000 pop.
Caffeine	-	-
Azithromycin	-	+ (142 ng/L)
Levothyroxine	+	+ (940 ng/L)
Omeprazole	-	+ (133 ng/L)
Fluoxetine	-	_



- ❖ Analytical method is sensitive. LODs ≈ 1 ng for most of the targeted analytes in the LCQ ZOOMTM mode.
- ❖ Method uses "green chemistry": < 5 mL of mobile phase per day; < 40 mLs solvent per extraction.</p>
- ❖ Certain of the targeted pharmaceuticals are present, at low levels, in wastewater effluent (azithromycin, levothyroxine, omeprazole).
- ❖ Many other untargeted, non-volatile and polar compounds are possibly present in wastewater effluent (e.g., PPGs and PEGs).



Environmental Sciences Division, EPA-Las Vegas

- ❖ Need to improve chromatography addition of ion pairing reagent, e.g., ammonium acetate
- Need to improve extraction method
- * Sample more WWTPs.
- Compare concentrations between influent and effluent of WWTPs.



For Others – e.g., Toxicologists, Engineers, etc.

- * Need to assess risk for both Humans and Wildlife from occurrence/exposure data as it is gathered by various scientists.
- * If exposure is a risk to environmental health, then need to develop better WWTP technologies to lower the risk.
- * Need to assess groundwater recharge issues.

